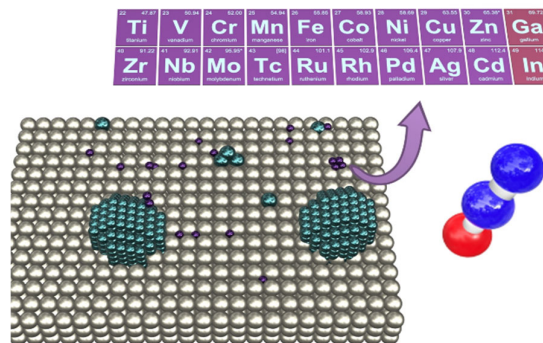


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Bimetallic catalytic systems for decomposition of nitrous oxide (N₂O)

Ammonia (NH₃) holds significant potential as a fuel for the shipping industry due to its favorable chemical properties. However, the emissions of gases such as N₂O, which has a GWP value 298 times higher than CO₂, pose a significant obstacle for the utilization of NH₃ as a climate-friendly fuel. A particular challenge involves the low-temperature abatement of N₂O and the inhibition effect produced by H₂O and O₂. Extensive research

has indicated that catalysts containing rhodium (Rh) are the most active in N₂O decomposition. Nevertheless, the role of the support and promoters, the use of bimetallic formulations, and the influence of the preparation method still need investigation. The effect of adding another metal and its influence on metal speciation is the particular interest as it could boost activity and stability.



The aim of this project is to synthesize a platform of bimetallic Rh-based catalysts, using different approaches to control the speciation and local structure of different metals and evaluate their performance for the decomposition of N₂O under realistic conditions (O₂ and H₂O). These results will guide efforts in the design of superior catalytic systems.

Tasks

- Synthesis of distinct matrix of catalysts based on ZrO_x and Rh.
- Characterization of synthesized materials using a variety of techniques (e.g. SEM, UV, H₂-TPR, O₂ chemisorption, N₂ physisorption, PXRD, IR of CO adsorption, TEM)
- Perform catalytic activity test
- Perform lab-based *operando* experiments

Benefits

- Hands-on approach in the catalysis field: from synthesis and characterization to testing and understanding
- Gaining experience on spectroscopic *operando* techniques
- Gaining experience on writing scientific reports/presentations
- Work in international team

The work will be carried out in the Applied Catalysis and Spectroscopy group at the Paul Scherrer Institut (PSI) in Villigen.

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