

Almond Shell-Derived Carbonaceous Materials as Anodes for Sodium-Ion Batteries

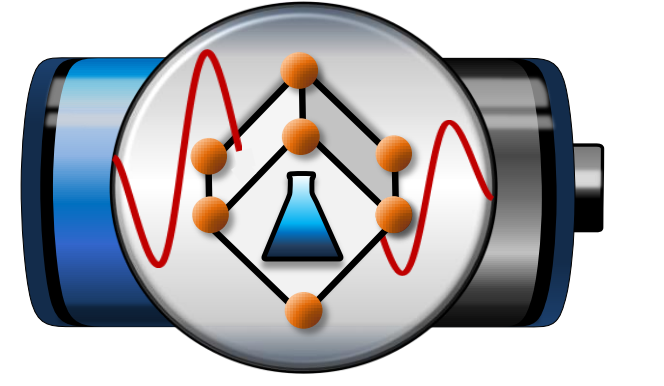


Swiss Competence Centers
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Objective

- To synthesize disordered carbon from biowaste materials
- To optimize synthesis procedure by varying the type of gas [Ar, N₂, Ar + 5% H₂ (AW5)] during synthesis
- To investigate their electrochemical performance as anode in Na-ion battery

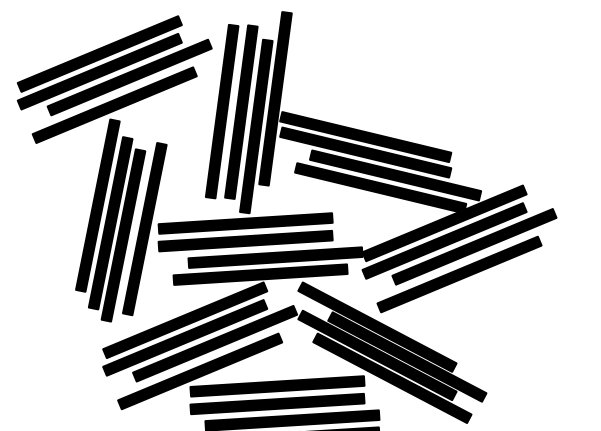
Synthesis



Washing
with 10%
(v/v) H₂SO₄

Washing with
distilled water
until neutral,
then drying at
80°C

Ball milling, then
carbonization at
1000°C under
different
atmospheres

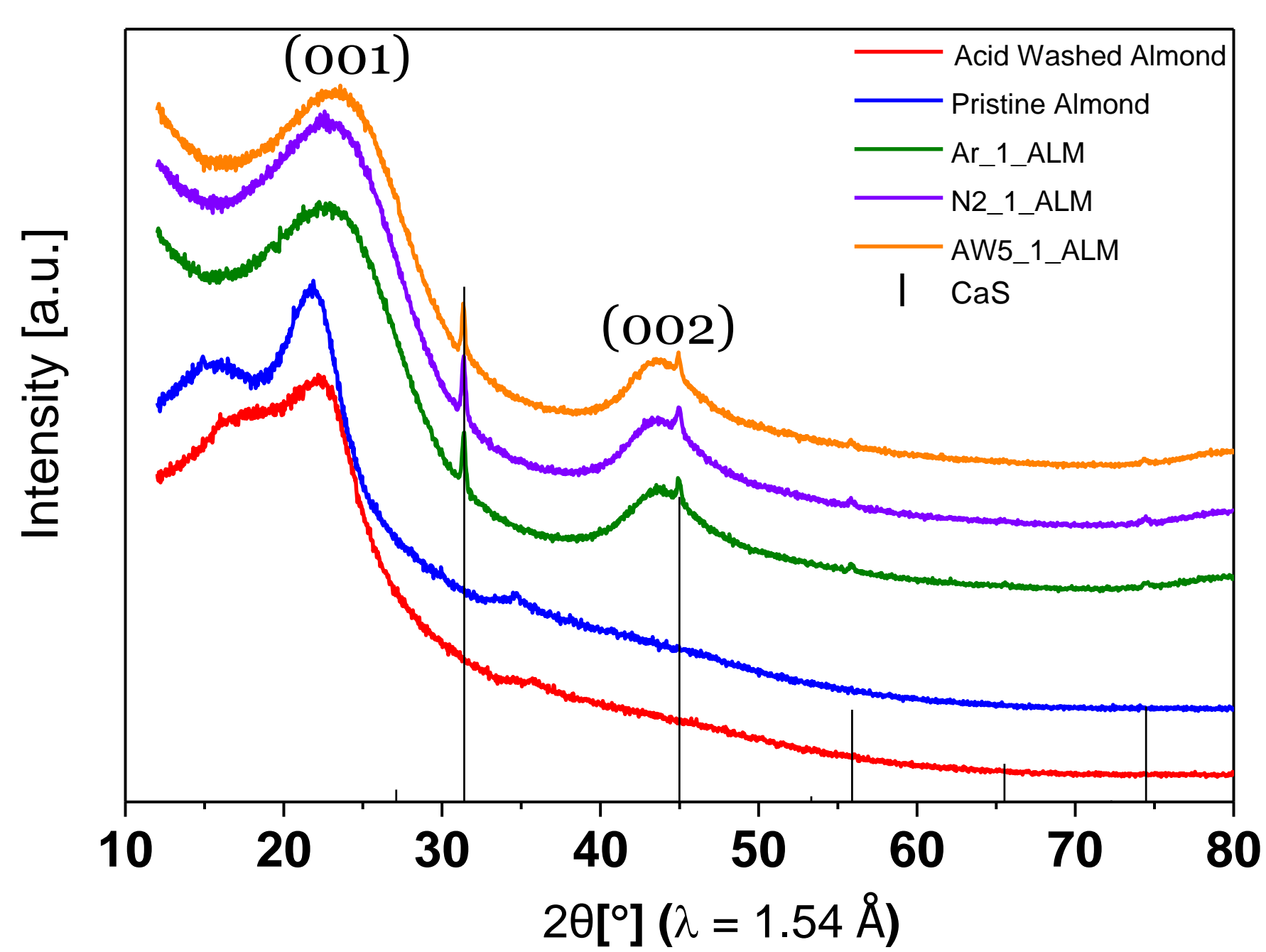


Disordered
carbon¹

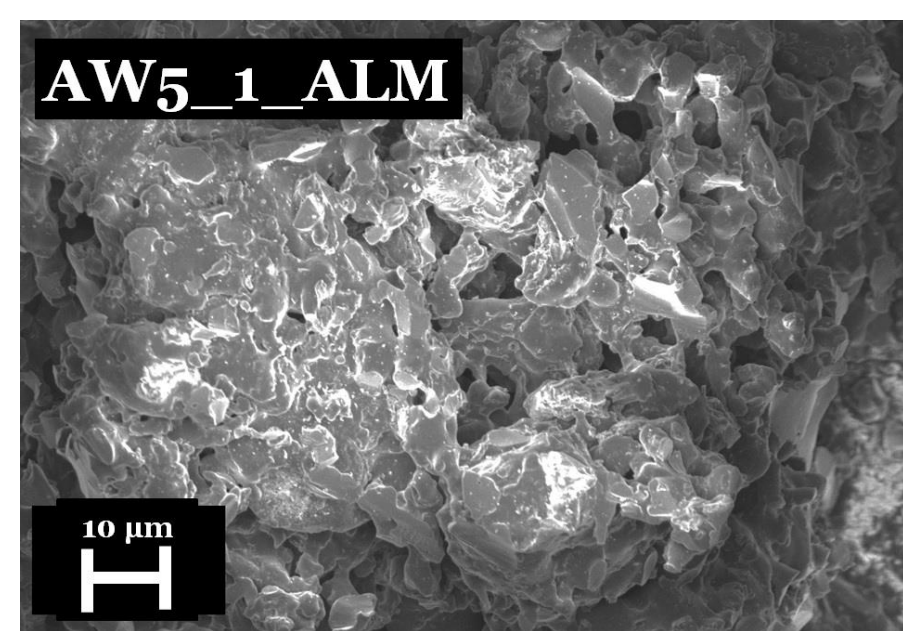
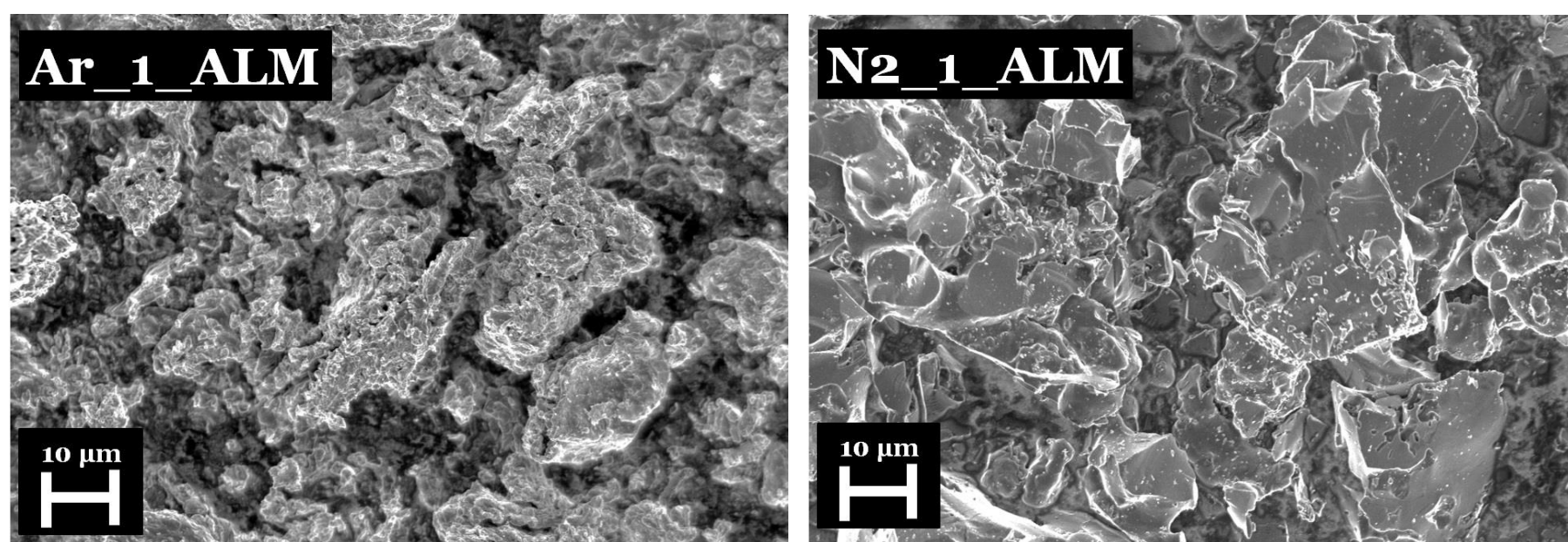
¹D. A. Stevens and J. R. Dahn, *Journal of the Electrochemical Society*, 2000.

Characterization Techniques

XRD and SEM

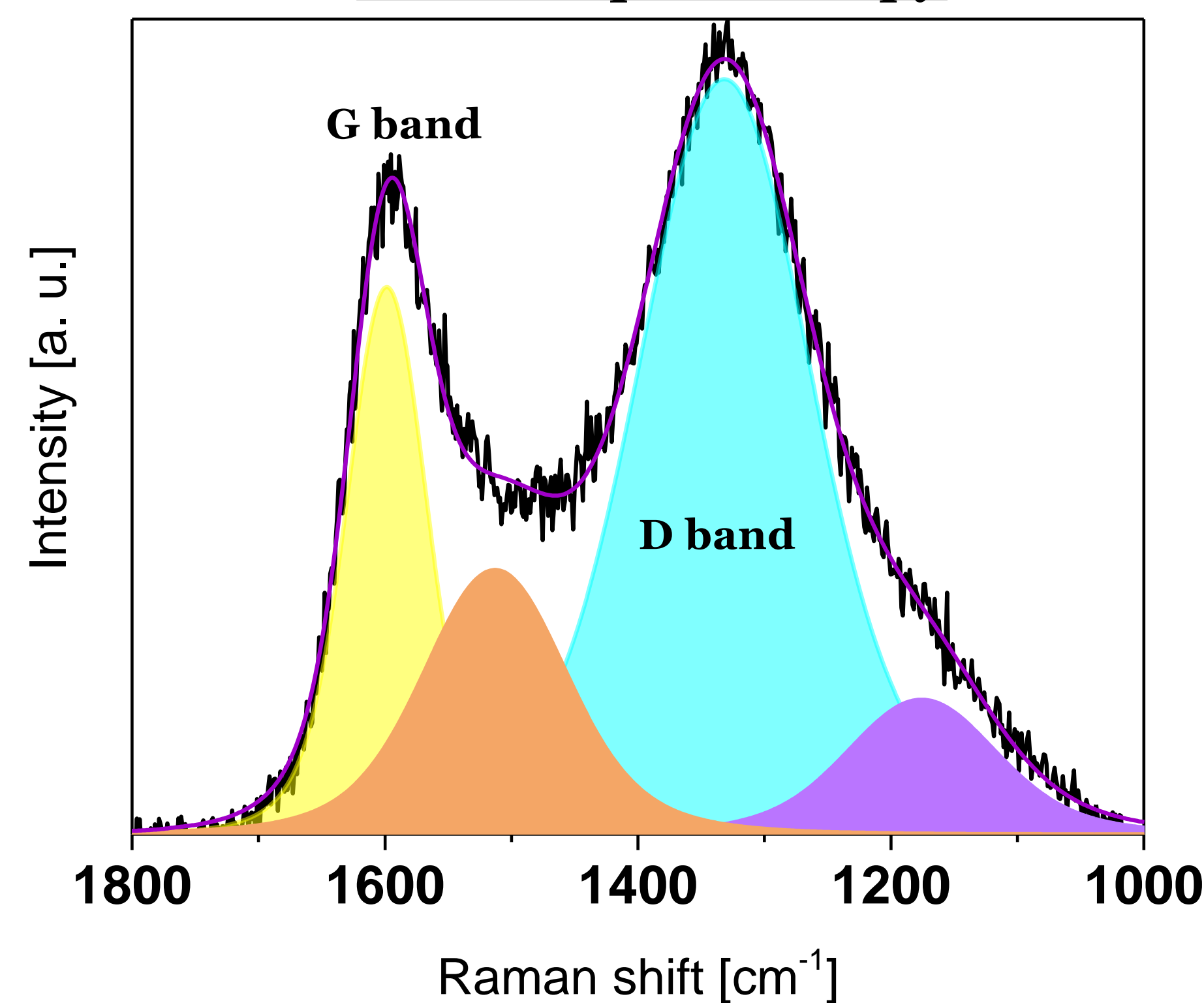


- ✓ CaS peaks appear due to reduction of CaSO₄ in almonds during carbonization



- ✓ Randomly shaped agglomerated carbon particles

Raman Spectroscopy



- ✓ Bands at ~1500 and ~1200 cm⁻¹ are due to sp³ and tetrahedral type carbons²
- ✓ Low I_D/I_G ratio, high degree of graphitization³

Samples	Raman shift (cm ⁻¹)		I _D /I _G ratio
	1 (G band)	3 (D band)	
Ar_1_ALM	1600	1331	1.38
N2_1_ALM	1600	1331	1.34
AW5_1_ALM	1600	1328	1.29

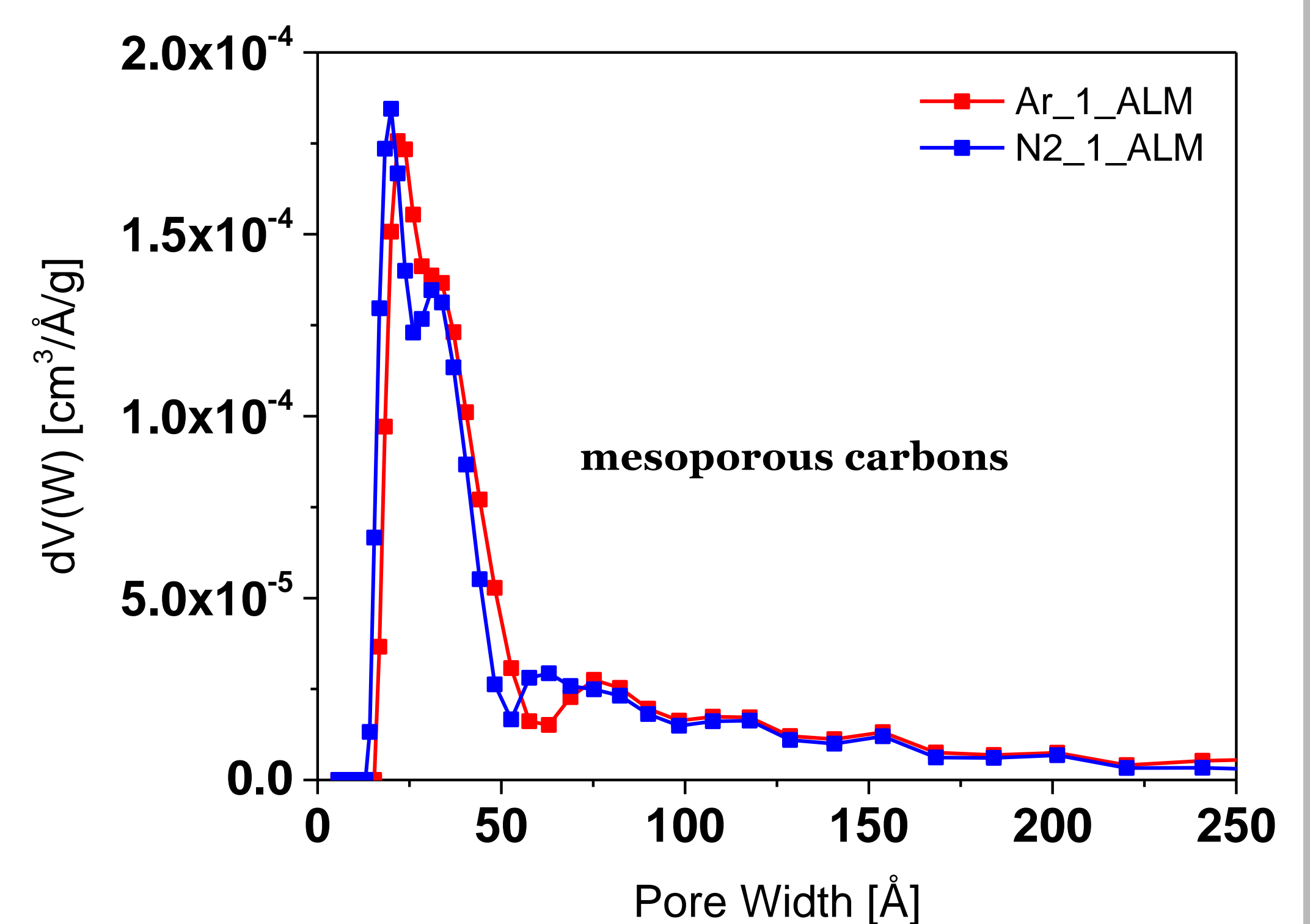
²M. Doeff, et al., *Electrochemical and Solid State Letters*, 2003.

³E. Irisarri et al., *Journal of the Electrochemical Society*, Review, 2015

BET Surface Area

Samples	Surface Area (m ² /g)	Total Pore Volume (cm ³ /g)*
Ar_1_ALM	4	8x10⁻³
N2_1_ALM	4	8x10⁻³

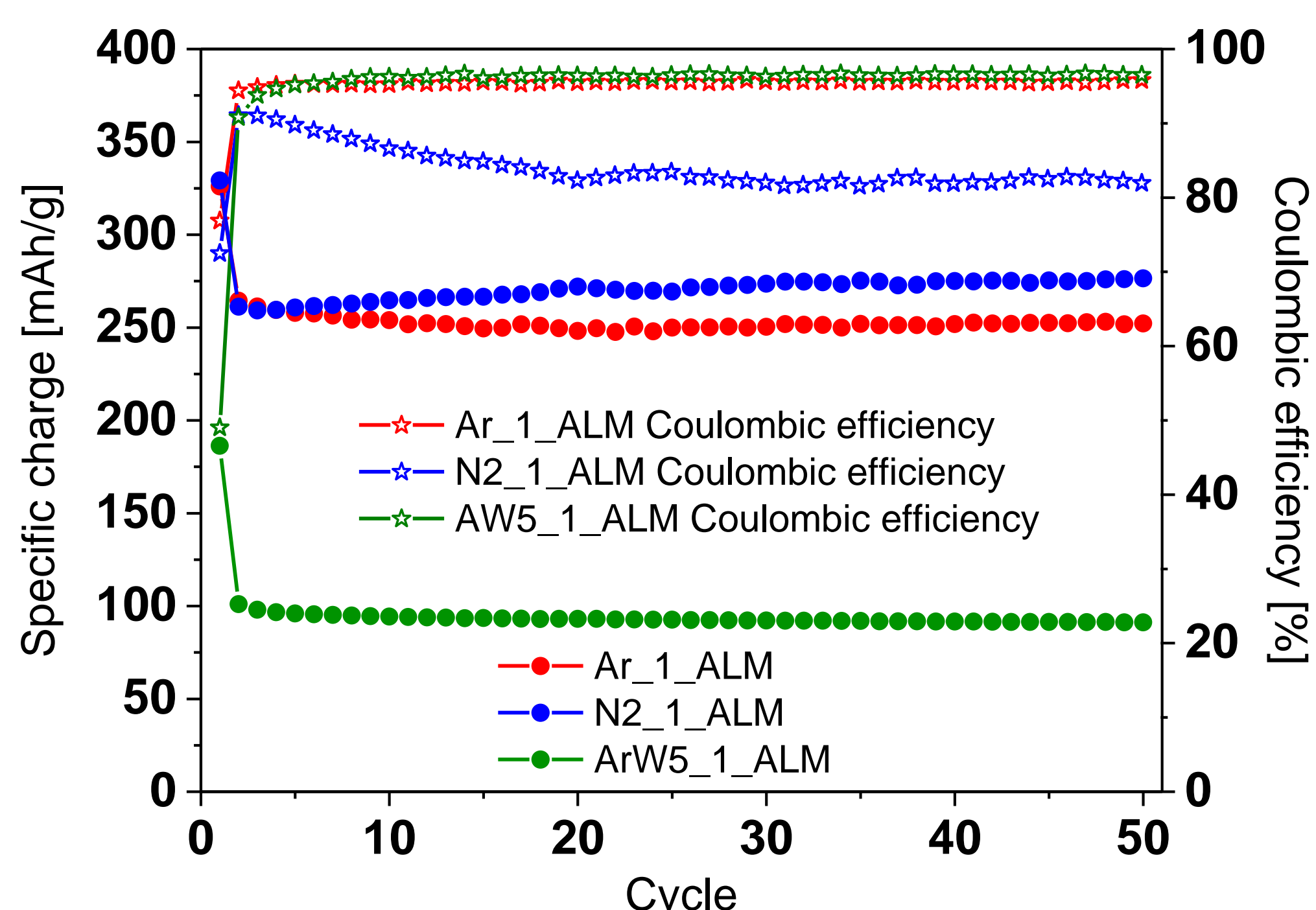
*at P/P₀ = 0.98897



- ✓ Similar surface area, total pore volume and pore size distribution for carbons synthesized under Ar and N₂

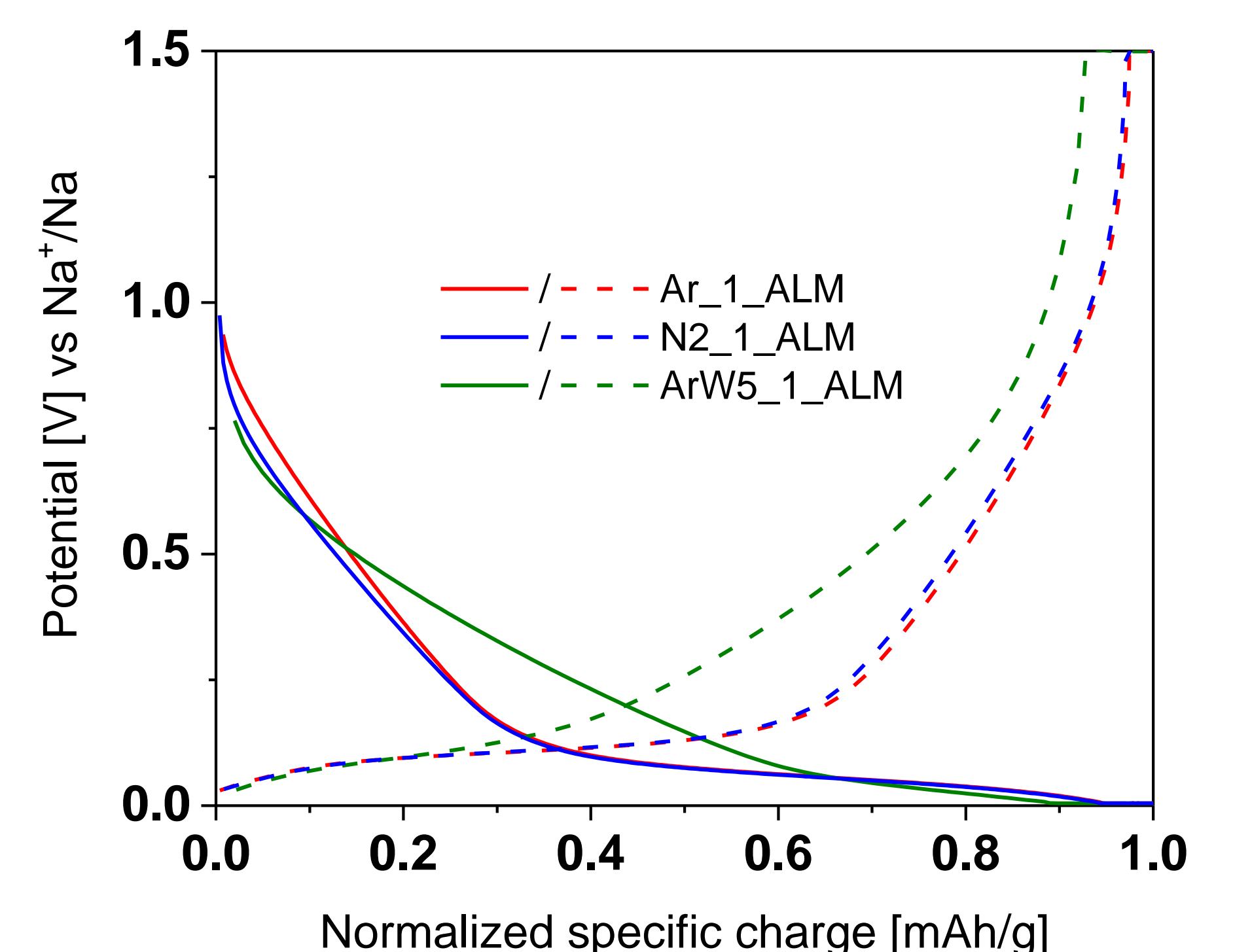
Electrochemical Performance

Cycling Performance and Galvanostatic Profiles



Electrolyte: 1 M NaClO₄ in PC
Cycling rate: C/10
Voltage window: 5 mV to 1.5 V

- ✓ Identical galvanostatic profiles and initial specific charge for Ar_1_ALM and N2_1_ALM
- ✓ Increasing specific charge for N2_1_ALM possibly due to electrolyte decomposition



Conclusions

- ✓ Disordered carbon anodes were successfully synthesized from a biowaste material
- ✓ Synthesis under a reductive atmosphere (AW5) led to a carbon anode with poor electrochemical performance
- ✓ Synthesis under Ar and N₂ led to carbonaceous materials with the same physical properties

Acknowledgment

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